Serbian Chemical Society Serbian Young Chemists' Club

Eight Conference of the Young Chemists of Serbia Book of Abstracts

Belgrade29th OCTOBER 2022

8th Conference of Young Chemists of Serbia Book of Abstracts

29th October 2022 University of Belgrade, Faculty of Chemistry 8th Conference of Young Chemists of Serbia Belgrade, 29th October 2022 Book of Abstracts

Published and organized by Serbian Chemical Society and Serbian Young Chemists' Club Karnegijeva 4/III, 11000 Belgrade, Serbia Tel./fax: +381 11 3370 467; www.shd.org.rs; office@shd.org.rs

Publisher Dušan SLADIĆ, president of Serbian Chemical Society

Editors

Jelena MILOVANOVIĆ Marko RODIĆ Vuk FILIPOVIĆ Života SELAKOVIĆ Jelena KESIĆ Mila LAZOVIĆ Mihajlo JAKANOVSKI

Page Layout and Design Vuk FILIPOVIĆ Jelena KESIĆ

Mila LAZOVIĆ Mihajlo JAKANOVSKI

Circulation 20 copies

ISBN 978-86-7132-080-1

Printing Development and Research Centre of Graphic Engineering Faculty of Technology and Metallurgy, Karnegijeva 4, Belgrade, Serbia

Scientific Committee

Dr. Jelena Milovanović – University of Belgrade, Institute of molecular genetics and genetic engineering

Dr. Marko Rodić - University of Novi Sad, Faculty of Sciences

Dr. Vuk Filipović – University of Belgrade, Institute of Chemistry, Technology and Metallurgy, National Institute of the Republic of Serbia

Dr. Života Selaković - University of Belgrade, Faculty of Chemistry

Organizing Committee

Jelena Kesić - University of Novi Sad, Faculty of Sciences

Mila Lazović - Innovative Centre of the Faculty of Chemistry, Belgrade

Mihajlo Jakanovski - Innovative Centre of the Faculty of Chemistry, Belgrade

European Young Chemists' Network

Dr. Maximillian Menche, chair of the EYCN

Sponsorship

The organizing committee is grateful for the donations of the selected sponsor participants

European Young Chemists' Network

Analysis doo

Ministry of Education, Science and Technological Development, Republic of Serbia







Republic of Serbia Minisrty of Education, Science and Technological Development

Acknowledgement

Acknowledgement to the University of Belgrade, Faculty of Chemistry for the use of the space of the Faculty during the 8th Conference of Young Chemists' of Serbia.

Thanks to the Serbian chemical society for the supporting during organization of the Conference.

Deeply acknowledgments to the European Young Chemists' Network and European Chemical Society for the financial support of the best oral and poster presentations.

Thanks to the Analysis d.o.o. for support and the promoting material.

Contents

Plenary Lecture	1
Invited Lectures	5
Oral communications	9
Poster presentations	21
Analytical chemistry	23
Biochemistry and biotechnology	51
Chemical education and history of chemistry	61
Chemistry of macromolecules and nanotechnology	63
Green chemistry	67
Inorganic chemistry	71
Material sciences	83
Medicinal chemistry	97
Organic chemistry	107
Physical chemistry	121
Theoretical chemistry	133
Author index	141

Belgrade, 29th October 2022

Organic chemistry

Supramolecular architectures of selected xanthenedione derivatives

<u>Anita M. Lazić¹</u>, Lidija D. Radovanović¹, Jelena R. Rogan², Nemanja P. Trišović² ¹ Innovation Center of Faculty of Technology and Metallurgy in Belgrade, Serbia ² University of Belgrade, Faculty of Technology and Metallurgy, Belgrade, Serbia

The wide range of pharmacological activities (e.g. antiviral, antifungal, antibacterical, antiinflamatory, leishmanicidal and antidepresant) has already been attributed to the xanthenediones, a group of synthetic heterocyclic compounds possessing a pyran nucleus fused on either side with cyclohex-2-enone rings [1]. In this work, two 3,3,6,6tetramethyl-9-substituted-3,4,5,6,7,9-hexahydro-1H-xanthene-1,8(2H)-diones (Figure 1) were synthesized and their crystal stuctures were determined by single crystal X-ray diffraction. The main structural feature in compound **1** is a supramolecular chain along the *a*-axis formed by O4–H4···O2 hydrogen bond and C13–H13···O4 and Br1···Br2 interactions between the adjacent asymmetric units, while the formation of supramolecular network is further achieved by $C-H\cdots\pi$ interactions between the adjacent chains. The main motif in 2 is a dimer formed via O4-H4...O2 hydrogen bond and Cl1... π interactions. The neighbouring dimers are connected through strong C7– H7A... π interactions, thus resulting in formation of a zigzag chain parallel to the *c*-axis. Weak C–H··· π interactions link the adjacent chains into a supramolecular layer. This work may provide a basis for design of new biologically active xanthenediones both at the molecular and supramolecular level.

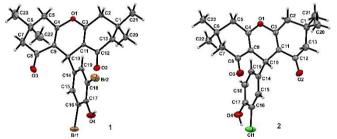


Figure 1. The asymmetric unit of 1 and 2 showing 30% displacement ellipsoids (the H atoms are presented as small spheres of arbitrary radii).

References

 I. E. Poly da Silva, M. Lopes da Silva, R. S. Dias, E. G. Santos, M. C. Brangioni de Paula, A. Silva de Oliveira, A. F. Costa da Silveira Oliveira, F. Marques de Oliveira, C. Canedo da Silva, R. R. Teixeira, S. Oliviera de Paula, *Microbes Infect.* 2020, 22(9), 489.

Acknowledgments

This work was supported by the Ministry of Education, Science and Technological Development of the Republic of Serbia (Contract No. 451-03-68/2022-14/200135; 451-03-68/2022-14/200287).

Supported by



Ministarstvo prosvete, nauke i tehnološkog razvoja Ministry of Education, Science and Technological Development





European Chemical Society —European Young Chemists' Network—

